

# School of Aeronautics (Neemrana)

**I-04, RIICO Industrial Area, Neemrana, Dist. Alwar, Rajasthan**

Approved by Director General of Civil Aviation, Govt. of India, All India Council for Technical Education  
Ministry of HRD, Govt of India & Affiliated to Rajasthan Technical University, Kota & BTU, Bikaner Rajasthan

Question Paper For Internal Assessment Examination (Theory) - Credit 3 / 31 / SET 1

### Instructions For Students / Faculty Mid Term I (Total 60 Marks, 2 HRS. Syllabus From Beginning Of Session)

• Part A: Total number of questions to be given are five, each carrying 3 marks and are compulsory to attend. There is no choice. They are short answer type questions (**Not More Than 25 Words For Both Question & Answer**), no objective type or fill in the blanks. Total 15 marks.

• Part B: Total number of questions to be given are six, out of which student has to answer any four. They are long answer type (**Not More Than 50 Words For Question**), each carrying 6 marks. Total 24 marks.

• Part C: Total number of questions to be given are four, out of which student has to answer any three. They are numerical answer type / fully elaborative type (**Not More Than 70 Words For Question**)\*, each carrying 7 marks. Total 21 marks.

### Mid Term II & III (Total 90 Marks, 2.5 HRS. Syllabus From Beginning Of Session)

• Part A: Total number of questions to be given are ten, each carrying 2 marks and are compulsory to attend. There is no choice. They are short answer type questions (**Not More Than 25 Words For Both Question & Answer**), no objective type or fill in the blanks. Total 20 marks

• Part B: Total number of questions to be given are seven, out of which student has to answer any five. They are long answer type (**Not More Than 50 Words For Question**), each carrying 6 marks. Total 30 marks.

• Part C: Total number of questions to be given are five, out of which student has to answer any four. They are numerical answer type / fully elaborative type (**Not More Than 70 Words For Question**)\*, each carrying 10 marks. Total 40 marks.

\* **LIST OF ELABORATIVE THEORY QUESTION SUBJECTS:** 3 MH4 - 07 Manufacturing Process, 4 AN4 - 06 Aircraft Materials and Processes (Cr 3), 5 AN4 - 05 Aircraft System (Cr 3), 6 AN4 - 05 Avionics-I (Cr 3), 6 MH4 - 03 Applied Hydraulics & Pneumatics (Cr 3), 6 MH5 - 11 Principles of Management (Cr 3), 6 MH5 - 13 Aircraft Electronics System (Cr 3), 7 AN5 - 12 Maintenance of Airframe and System (Cr 3), 7 AN5 - 13 Helicopter Theory (Cr 3), 7 AG6 - 60.1 Human Engineering and Safety (Cr 3), 7 ST - 01 Avionics II (Special Theory Subject) (Cr 3), 7 MH5 - 11 Design of Mechatronics Systems (Cr 3), 7 MH5 - 12 Robotics and Machine Vision System (Cr 3), 7 MH6 - 13 Medical Electronics (Cr 3), 7 AN6 - 60.1 Aircraft Avionic System (Cr 3), 8 AN5 - 12 Maintenance of Power Plant and System (Cr 3), 8 AN5 - 13 Unmanned Aerial Vehicles & Systems (UAV) (Cr 3), 8 MH5 - 13 Product Development & Launching (Cr 3), 8 EC6 - 60.2 Robotics and control (Cr 3)

FACULTY MEMBERS, PLEASE ENSURE EXCEPT ABOVE LISTED SUBJECTS, NO THEORITICAL ELABORATIVE QUESTION SHOULD BE GIVEN IN PART 'C' OF QUESTION PAPER

### Question Paper & Student Details

<b>Mid Term</b>	Mid Term 2	<b>Date of Submission</b>	21/08/2020
<b>Name of Faculty</b>	Mr. Sidhartha Sondh	<b>Date of Examination</b>	25/08/2020
<b>Course</b>	B.Tech (Aeronautical Engineering)	<b>Semester</b>	SEMESTER : 3
<b>Batch</b>	Combined Batches 18, 19, SF 2	<b>Subject</b>	3 AN3 - 03 Engineering Thermodynamics (Cr 3)

### COURSE OUTCOMES FOR REFERENCE TO FRAME QUESTION PAPER

*(Faculties are required to mention relevant Course Outcome number against the respective question in QP)*

<b>Course Outcome</b>	COURSE OUTCOME Upon completion of the course, Students will be able to CO1: Apply different laws and basics of thermodynamic process. CO2: Analyze cyclic & Non-cyclic processes using Zeroth and First law of thermodynamics. CO3: Relate theoretical knowledge with various flow processes like steady flow, unsteady flow. CO4: Interpret the second law of thermodynamics for various engineering systems.		
<b>Email I'd</b>	sidharthasondh@soaneemrana.org	<b>Phone No.</b>	963-455-7511
<b>Student Name</b>		<b>Student Reg No.</b>	

<b>Part A</b>			
<b>Question : 1</b>	What are the different types of system?		
1	Fundamentals	Self	1
<b>Question : 2</b>	What is the necessary condition to determine the entropy of a series of process?		
16	Entropy	Self	4

<b>Question : 3</b>	What is the significance of Carnot efficiency?		
14	Carnot engine	Self	2
<b>Question : 4</b>	What is a throttling process?		
13	Application of 1st law	Self	2
<b>Question : 5</b>	Differentiate between steady flow process and unsteady flow process.		
14	Application of 1st law	Self	2
<b>Question : 6</b>	What s a control volume?		
16	Control volume	Self	2
<b>Question : 7</b>	Define Carnot engine.		
14	Carnot engine	Self	4
<b>Question : 8</b>	What is T-s Diagram?		
10	Property	Self	2
<b>Question : 9</b>	Is it possible to mathematically prove second law of thermodynamics? Give reason.		
16	Second law of thermodynamics	Self	4
<b>Question : 10</b>	State Clausius statement of second law of thermodynamics.		
16	Second law of thermodynamics	Self	3
<b>Part B</b>			
<b>Question : 1</b>	Justify the statement "Entropy is degradation of energy". Support your answer with a valid example.		
18	Entropy	Self	3
<b>Question : 2</b>	Define COP. An inventor claims to have developed a refrigerator that maintains the refrigerated space at 1°C while operating in a room where the temperature is 24°C and that has a COP of 13.5. Is this claim reasonable? Support your answer with proper reason.		
16	Heat Pump	Engineering Thermodynamics By Cengel and Boles	4
<b>Question : 3</b>	Describe the thermodynamic temperature scale. How is it different from other temperature scale.		
8	Temperature scale	Self	2
<b>Question : 4</b>	Explain what is irreversibility? What are the different types of irreversibility that occur during a process?		
16	Irreversibility	Self	4
<b>Question : 5</b>	Perform energy analysis for Carnot cycle.		
14	Carnot cycle	Self	4
<b>Question : 6</b>	Is it possible for a heat engine to operate without rejecting any waste heat to a low temperature reservoir? Explain.		
14	Heat engine	Self	4
<b>Question : 7</b>	Differentiate between the following: (i) Heat pump and refrigerator (ii) Efficiency and COP.		
15	Heat pump	Self	3
<b>Part C</b>			
<b>Question : 1</b>	Air at 10°C and 80 kPa enters the diffuser of a jet engine steadily with a velocity of 200 m/s. The diffuser requires a heat of 20 kJ. The air leaves the diffuser with a velocity that is 10 m/s producing a work of 500 W. Determine (a) the mass flow rate of the air and (b) the temperature of the air leaving the diffuser.		

8	Application of First law	Engineering Thermodynamics By Cengel and Boles	3
<b>Question : 2</b>	Steam enters a nozzle at 400°C and 800 kPa with a velocity of 10 m/s, and leaves at 300°C and 200 kPa while losing heat at a rate of 25 kW. For an inlet area of 800 sq. cm, determine the velocity and the volume flow rate of the steam at the nozzle exit.		
9	Nozzle	Engineering Thermodynamics By Cengel and Boles	3
<b>Question : 3</b>	Derive expression for filling water in a tank.		
12	Unsteady flow	Self	3
<b>Question : 4</b>	A heat pump is used to meet the heating requirements of a house and maintain it at 20°C. On a day when the outdoor air temperature drops to -2°C, house is estimated to lose heat at a rate of 80,000 kJ/h. If heat pump under these conditions has COP of 2.5, determine (a) power consumed by the heat pump (b) rate at which heat is absorbed from cold outdoor air.		
16	Heat Pump	Engineering Thermodynamics By Cengel and Boles	4
<b>Question : 5</b>	Consider a steady-flow heat exchanger involving two different fluid streams. Under what conditions will the amount of heat lost by one fluid be equal to the amount of heat gained by the other? Explain.		
9	Application of 1st law	Self	3
<b>Upload Scanned Document In Case of Numerical or Diagram For Any of The Above Questions.</b> (Mention question number with relevant fig / numerical / equations. Max 150 KB)			
<b>I have scrutinized the question paper. There is no spelling mistake or any type of irrelevant question.</b>		SS	

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